Particle size of aggregate and its influence upon the expansion caused by the alkali-silica reaction*

D. W. Hobbs and W. H. Gutteridge

Contribution by Alek Samarin

Ready Mixed Concrete Ltd, Australia, Central Research Laboratories

One of the most effective methods of inhibiting the expansion of reactive aggregate is by the addition of a sufficiently large quantity of pozzolana to concrete. The method is well publicized and documented, and even covered by ASTM Designation C441-69 (re-approved in 1975). "Standard test method for effectiveness of mineral admixtures of concrete due to the alkali-aggregate reaction".

Pozzolans produced by fine-grinding of potentially reactive opaline materials⁽¹⁾ and even pozzolans with relatively high alkali contents⁽²⁾ have been reported to be quite effective inhibitors of alkali-aggregate reaction. Hence, one of the potential methods of prevention of alkali-aggregate reaction may be by means of pulverization of the portion of reactive aggregate in lieu of additions of pozzolana from external sources, when the latter are not economically available in a particular location.

Potential pozzolanicity of pulverized material can be evaluated either by the ASTM method mentioned

above or by means of X-ray diffraction (XRD) analysis. For example, pulverized beer-bottle glass⁽³⁾ was found to be pozzolanic, as shown in Figure I (overleaf).

According to Celani et al.⁽⁴⁾, in mixes containing pozzolana no line at $9.80 \text{ \AA}$ (which is normally associated with calcium silicate hydrate of high CaO/SiO_2 ratio) is present. Portlandite is characterized by the distinct lines at 4.90 , 2.628 , 1.927 and $1.796 \text{ \AA}$.

The differences between XRD read-out on a Portland cement matrix and a matrix containing pulverized glass do confirm the pozzolanic character of fine ground glass. In contrast, large particles of the same crushed glass exhibited typical alkali-aggregate reaction⁽³⁾. Utilization of pozzolanic properties of finely ground reactive materials may well be one of the practical aspects, related to the influence of particle size upon the expansion caused by the alkali-silica reaction.

Reply by the authors

We thank Mr Samarin for his interest in our paper and agree that the addition of finely ground reactive material might be an effective method of reducing expansion due to the alkali-silica reaction. The amount of ground material required is dependent upon the alkali content of the cement and the intended mix proportions of the concrete. We suggest

that this pulverized material reacts rapidly with the water-soluble alkalis from the cement and that no deleterious expansion will occur when the water-soluble alkali level in the concrete is reduced to less than 2.5 kg/m^3 ⁽⁵⁾. We deduce from Diamond and Thaulow's data⁽⁶⁾ that the reactive material must be ground to less than $20 \text{ }\mu\text{m}$ for it to be effective.

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(a) Portland cement matrix

(b) Portland cement matrix containing beer-bottle glass
Figure 1: CuK α radiation results.

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